



A Course of Instruction and System of Procedure in Qualitativ Chemical Analysis

By Arthur Amos Noyes

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(potassium cadmicyanide), since its complex anion has a much smaller ionization tendency than the cation $Cd(NH_3)_4^{++}$; thus in a normal KCN solution the ratio of the concentration of the complex anion to that of the simple Cd^{++} ion is about 1017. Yet this complex anion is sufficiently dissociated into Cd^{++} to cause the solution to become supersaturated with CdS when an alkaline sulfide is added. 4. A very small black precipitate (which may be due to HgS or PbS) may sometimes be produced in the final test for cadmium with H_2S ; but, provided the analysis has been properly conducted, not in sufficient quantity to prevent the yellow color of 1 or 2 mg. of CdS from being seen. In case a black precipitate is produced, and (hus prevents a positive conclusion as to the presence or absence of a little cadmium, the precipitate may be treated,...

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